SIDDHARTH INSTITUTE OF ENGINEERING & TECHNOLOGY:: PUTTUR

(AUTONOMOUS)

B.Tech I Year I Semester Regular & Supplementary Examinations March-2023 APPLIED CHEMISTRY

(Common for EEE & ECE)

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	Т	Fime: 3 hours	Max. Marks: 60		
		(Answer all Five Units $5 \times 12 = 60$ Marks) UNIT-I			
1	a	Define Fuel cell? Describe the Construction and Working principle and uses of Methanol – Oxygen Fuel cell.	CO1	L1	6M
	b	Write a note on Lithium-Ion rechargeable cell. OR	CO1	L2	6M
2	a		CO1	L3	6M
	b	Explain about Potentiometric redox titrations. UNIT-II	CO1	L2	6M
3	a	Write the postulates of molecular orbital theory.	CO ₂	L2	6M
3	b		CO ₂	L2 L2	6M
		OR			
4	a	Explain colour properties of transition metal complexes.	CO ₂	L2	6M
		Explain the crystal field splitting in Tetrahedral complexes. UNIT-III	CO2	L3	6 M
5	a	Explain the free radical addition polymerization.	CO ₃	L2	6M
	b		CO3	L4	6M
		OR			
6	a	Write the preparation, properties and application of Buna-S rubber and Buna-N rubber.	CO3	L2	8M
	b	What are conducting polymers? How are they classified? UNIT-IV	CO3	L1	4M
7	a	Explain the principle, working and applications of Thin Layer Chromatography (TLC).	CO4	L2	6M
	b	Write the applications of IR spectroscopy. OR	CO4	L2	6M
8	a	Explain principle & instrumentation of UV-visible spectroscopy with neat diagram.	CO4	L2	9M
	b	Mention any six applications of UV-Visible spectroscopy. UNIT-V	CO4	L2	3M
9		What is meant by Nano materials? How the Nano materials Classified.	CO ₅	L1	6M
	b	Write a short notes on Carbon Nano Tubes.	CO ₅	L1	6M
		OR			
10		Discuss about the principle and application of Super conductors and their applications.	CO5	L1	10M
	b	Define Super conductors.	CO5	L1	2M

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STORIL GRADIES THE OF ENGINEERING & TECHNOLOGY: PUT FUR

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			Write the preparation, properties and application of Buna-S subber and Buna-N subber.					